

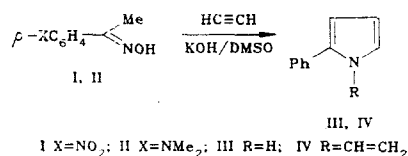
UNEXPECTED LOSS OF FUNCTIONALITY DURING THE TROFIMOV SYNTHESIS
OF PYRROLES FROM 4-NITRO AND 4-DIMETHYLAMINOACETOPHENONE OXIMES

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UDC 547.741

Condensation of ketoximes with acetylene in highly basic media primarily gives rise to pyrroles and N-vinylpyrroles [1]. It has been reported [2] that 4-nitroacetophenone oxime (I) in KOH-DMSO does not give the corresponding pyrroles with acetylene.

In continuation of attempts to synthesize 2-(4-nitrophenyl)pyrroles we have found that condensation of oxime I with acetylene occurs with loss of the NO₂ group and formation of 2-phenylpyrrole (III, yield ~7%) and 1-vinyl-2-phenylpyrrole (IV, yield 1.4%). The reaction was carried out for 3 h at 100°C with a molar ratio of I:KOH of 1:2 using an autoclave and an initial acetylene pressure of 14 atmospheres.



A similar reaction course was observed when the dimethylamino analog (II) was condensed with acetylene at atmospheric pressure (5 h, molar ratio of II:KOH = 1:3, 100°C). Along with the expected NH and N-vinylpyrroles (2 and 4% yields correspondingly as determined by GLC but with purity insufficient for determination of physical constants) the pyrroles III and IV were also formed in overall 4% yield.

Pyrroles III and IV were separated chromatographically from the reaction mixture on a thin, unbound layer of aluminum oxide using hexane-ether (1:1) as eluent. GLC showed them to have the same retention time as reference samples and their spectra (IR, ¹H and ¹³C NMR) were identical with the samples prepared by condensation of acetophenone oxime with acetylene [3, 4].

LITERATURE CITED

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